

Structural Insights into DNA Replication without Hydrogen Bonds

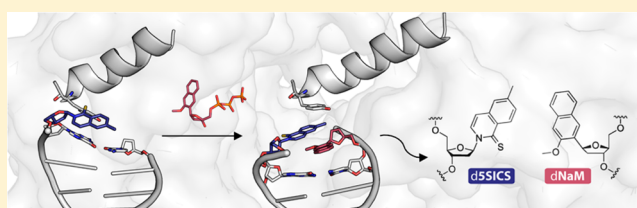
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S Supporting Information

ABSTRACT: The genetic alphabet is composed of two base pairs, and the development of a third, unnatural base pair would increase the genetic and chemical potential of DNA. d5SICS-dNaM is one of the most efficiently replicated unnatural base pairs identified to date, but its pairing is mediated by only hydrophobic and packing forces, and in free duplex DNA it forms a cross-strand intercalated structure that makes its efficient replication difficult to understand. Recent studies of the KlenTaq DNA polymerase revealed that the insertion of d5SICSTP opposite dNaM proceeds via a mutually induced-fit mechanism, where the presence of the triphosphate induces the polymerase to form the catalytically competent closed structure, which in turn induces the pairing nucleotides of the developing unnatural base pair to adopt a planar Watson–Crick-like structure. To understand the remaining steps of replication, we now report the characterization of the prechemistry complexes corresponding to the insertion of dNaMTP opposite d5SICS, as well as multiple postchemistry complexes in which the already formed unnatural base pair is positioned at the postinsertion site. Unlike with the insertion of d5SICSTP opposite dNaM, addition of dNaMTP does not fully induce the formation of the catalytically competent closed state. The data also reveal that once synthesized and translocated to the postinsertion position, the unnatural nucleobases again intercalate. Two modes of intercalation are observed, depending on the nature of the flanking nucleotides, and are each stabilized by different interactions with the polymerase, and each appear to reduce the affinity with which the next correct triphosphate binds. Thus, continued primer extension is limited by deintercalation and rearrangements with the polymerase active site that are required to populate the catalytically active, triphosphate bound conformation.



INTRODUCTION

Successful development of a functional unnatural base pair that is orthogonally replicated in DNA is the first step toward creating a semisynthetic organism with increased potential for information storage and retrieval, and would also expand the utility of nucleic acids for biological and biotechnological applications.^{1–10} While a variety of unnatural base pair candidates have been reported,^{11–15} only three have been shown to be efficiently replicated,^{16–18} and only the pair formed between d5SICS and dNaM (d5SICS-dNaM; Figure 1) has been shown to be PCR amplified without sequence-bias¹⁹ and efficiently transcribed in both directions.^{20,21}

The efficient replication of d5SICS-dNaM is particularly interesting because it proceeds in the absence of complementary hydrogen bonds (H-bonds) that underlie Watson–Crick-like pairing, and indeed, it forms an intercalated structure in duplex DNA.^{22,23} This mode of pairing maximizes packing interactions, and is likely general for nucleotides with predominantly hydrophobic nucleobases,^{24,25} but the resulting structure is reminiscent of a mismatch between natural nucleotides^{26–31} and is thus difficult to reconcile with efficient polymerase recognition. To investigate the structural basis for the efficient replication of DNA containing d5SICS-dNaM, we

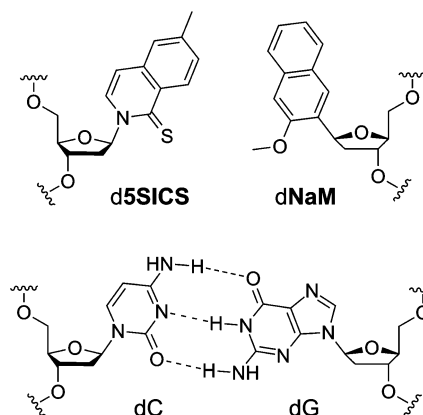


Figure 1. d5SICS-dNaM unnatural base pair, with a natural Watson–Crick base pair shown for comparison.

recently solved the crystal structure of KlenTaq DNA polymerase, the large fragment of the type I DNA polymerase from *Thermus aquaticus*, complexed with a templating dNaM,

Received: September 16, 2013

Published: November 27, 2013

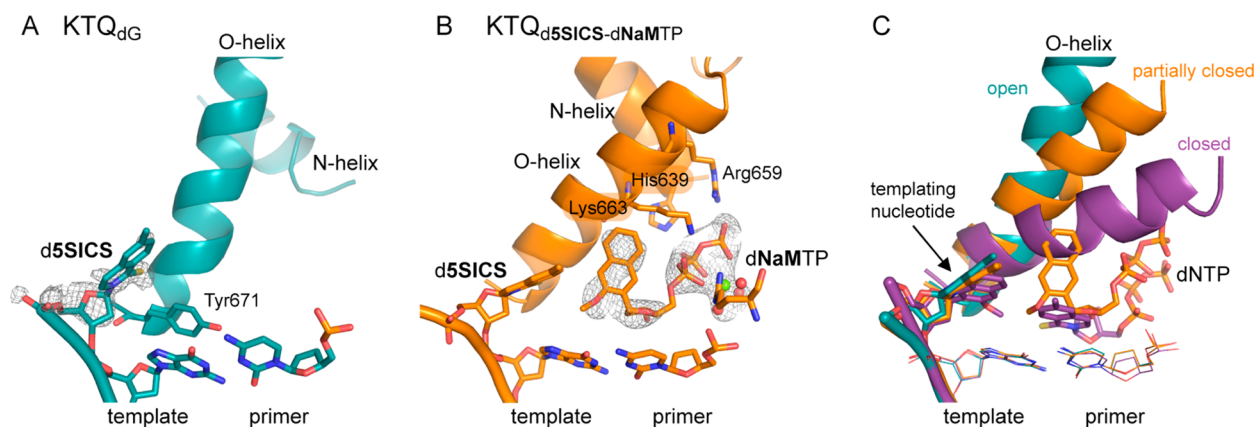


Figure 2. Open binary and precatalytic ternary complex of $\text{KTQ}_{\text{d5SICS}}$ and $\text{KTQ}_{\text{d5SICS-dNaMTP}}$, respectively. (A) The natural base pair at the postinsertion site, the templating d5SICS, and Tyr671 are shown as sticks, and the O- and N-helices are shown as cartoon. Simulated annealing mFo-DFc omit map around d5SICS is shown, contoured at 3σ . (B) Same arrangement as in (A) but for the ternary complex $\text{KTQ}_{\text{d5SICS-dNaMTP}}$. Simulated annealing mFo-DFc omit map around the bound dNaMTP and the coordinated Mg^{2+} ion (green sphere) and associated water molecules (red spheres) is shown, contoured at 3σ . (C) Superposition of $\text{KTQ}_{\text{d5SICS}}$ (cyan), $\text{KTQ}_{\text{d5SICS-dNaMTP}}$ (orange), and $\text{KTQ}_{\text{dNaM-d5SICS}}$ (PDB ID 3SZ2, purple) shows the open, partially closed, and closed state of the enzyme.

and with or without bound d5SICSTP ($\text{KTQ}_{\text{dNaM-d5SICSTP}}$ and KTQ_{dNaM} , respectively).²² The structures of these prechemistry d5SICSTP incorporation complexes revealed that the pairing of d5SICSTP with dNaM drives the open-to-closed conformational change characteristic of a natural base pair,^{32–34} and interestingly, once in the closed environment, the pairing unnatural nucleotides adopt a planar, Watson–Crick-like geometry.²² Thus, we demonstrated that not only is the polymerase able to select for pairs that form a correct Watson–Crick structure, but at least with hydrophobic analogues it is able to enforce the correct structure. This mutually induced fit mechanism highlights what might be a fundamental advantage of using hydrophobicity and packing forces to mediate replication, as they are sufficiently strong to mediate pairing, but also sufficiently plastic to adapt to the structure required by the polymerase. However, the insertion of d5SICSTP opposite dNaM is a particularly efficient step of replication,¹⁶ and the mechanism by which dNaMTP is inserted opposite d5SICS and the mechanism by which the primer containing either unnatural nucleotide is further elongated, which actually limits replication, remained unclear.

Here, to fully characterize the mechanism of unnatural base pair replication, we report the crystal structures of the prechemistry incorporation complexes leading to the insertion of dNaMTP opposite d5SICS: the binary complex of KlenTaq with a DNA template containing d5SICS at the templating position ($\text{KTQ}_{\text{d5SICS}}$) and the corresponding ternary complex with dNaMTP bound ($\text{KTQ}_{\text{d5SICS-dNaMTP}}$). We also report the structure of four postincorporation complexes: the binary complexes of KlenTaq and either a primer terminating with d5SICS paired opposite dNaM in a template ($\text{KTQ}_{\text{dNaM-d5SICS}}$) in three different sequence contexts, or a primer terminating with dNaM paired opposite d5SICS ($\text{KTQ}_{\text{d5SICS-dNaM}}$). Along with our previously reported structures, these structures provide key insights into the replication of the unnatural base pair and elucidate a mechanism that is based on a balance of intercalation and deintercalation and structural rearrangements of the polymerase active site.

RESULTS

Prechemistry dNaMTP Incorporation Complexes.

KlenTaq was first crystallized bound to a DNA primer/template with d5SICS at the templating position (position n). In the resulting complex ($\text{KTQ}_{\text{d5SICS}}$; Figure 2A), the d5SICS nucleoside adopts an extrahelical position that is similar to that observed for a natural dG in KTQ_{dG} (PDB ID 3SZ2).²² However, compared to the previously described binary structures KTQ_{dNaM} (PDB ID 3SYZ) and KTQ_{dT} (PDB ID 3SV4), the single-stranded DNA of the template adopts a different arrangement (Figure S1, Supporting Information). It is likely that the single-stranded portion of the template is flexible in the binary structures, and that the differences are not functionally relevant. Structural heterogeneity of the template is also implied by the absence of well-defined electron density for the d5SICS nucleobase.

To determine whether the addition of dNaMTP to $\text{KTQ}_{\text{d5SICS}}$ drives the same conformational change observed upon d5SICSTP binding to KTQ_{dNaM} , we next determined the structure of $\text{KTQ}_{\text{d5SICS-dNaMTP}}$ by soaking $\text{KTQ}_{\text{d5SICS}}$ crystals with dNaMTP. The structure of $\text{KTQ}_{\text{d5SICS-dNaMTP}}$ reveals that the unnatural triphosphate is bound to the O-helix (Figure 2B), which is rotated and only partially closed. The position of the triphosphate appears to be stabilized by ionic interactions with Arg659 and Lys663 of the O-helix, as well as with His639 of the N-helix and Arg587 from the N-terminal end of the thumb domain K-helix. In addition, along with three water molecules, the triphosphate moiety coordinates a Mg^{2+} ion. The electron densities for the sugar and the nucleobase moieties of dNaMTP are less well-defined than that for the triphosphate moiety, suggesting an increased level of disorder and/or flexibility. The N- and O-helices of the fingers domain adopt a conformation intermediate between the open and closed states (Figure 2C) (the root-mean-square deviation (rmsd) of residues 637–700 is 1.59 and 2.23 Å relative to $\text{KTQ}_{\text{d5SICS}}$ and $\text{KTQ}_{\text{dNaM-d5SICSTP}}$, respectively). In addition, Tyr671 is slightly displaced from its open conformation position in the insertion site (Figure S2), and the templating unnatural nucleobase moves from its extrahelical position toward the insertion site, again representative of a state intermediate between the open and closed conformations.

Postchemistry Extension Complexes. We next sought to investigate the structures of the postchemistry complexes, with the unnatural base pair positioned in the postinsertion site, where it is poised for continued primer elongation (i.e., extension of the unnatural base pair). We first characterized the structure of $\text{KTQ}_{\text{dNaM-d5SICS}}$ with d5SICS at the primer terminus paired opposite dNaM at the n-1 position, with three different primer/template complexes (E1–E3, Table 1).

Table 1. Primer/Template Sequences of Postchemistry Complexes Characterized

postchemistry complex	primer/template sequence	PDB ID
$\text{KTQ}(\text{E1})_{\text{dNaM-d5SICS}}$	5'-ACC ACG GCG C 5SICS 3'-TGG TGC CGC G NaM GA	4C8L
$\text{KTQ}(\text{E2})_{\text{dNaM-d5SICS}}$	5'-GCC ACG GCG C 5SICS 3'-CGG TGC CGC G NaM CTT	4C8O
$\text{KTQ}(\text{E2})_{\text{d5SICS-dNaM}}$	5'-GCC ACG GCG C NaM 3'-CGG TGC CGC G 5SICS CTT	4C8M
$\text{KTQ}(\text{E3})_{\text{dNaM-d5SICS}}$	5'-ACC ACG GCG C 5SICS 3'-TGG TGC CGC G NaM GTT	4C8N

In each binary complex, the polymerase adopts the expected open conformation, similar to that observed in KTQ_{dNaM} , $\text{KTQ}_{\text{d5SICS}}$, or KlenTaq bound to a fully natural primer/template.²² However, the presence of the unnatural base pair has a significant effect on the structure of the primer/template. In the structure of $\text{KTQ}(\text{E1})_{\text{dNaM-d5SICS}}$, the template dNaM cross-strand intercalates into the primer strand between d5SICS and the 5' dC ($\text{dC}_{\text{n-2}}$) (Figures 3 and S3). To accommodate this intercalation, relative to their positions observed with natural substrates, the C1' of the primer unnatural nucleotide moves 4.7 Å toward the template and the C1' of the unnatural nucleotide in the template shifts 4.5 Å in the direction of translocation (Figure 4A and C). The extent of intercalation is evident by the sugar C1'–C1' distance of 8.4 Å, compared to the ~10.5 Å distance that is typical for a natural pair in the postinsertion site.^{22,33} This degree of intercalation is even greater than in the free duplex, where the C1'–C1' distance is 9.1 Å,²² likely reflecting a decreased level of structural restraints when the unnatural base pair is positioned at the end of a duplex, as opposed to the middle. Intercalation also positions the templating nucleobase proximal to the primer terminus, and the N1 and C2 amino group of dG_{n} form H-bonds with the phosphate backbone of the primer (Figure S4).

Although perturbations are apparent with the n-2 and n-3 template nucleotides, they are smaller, and the remainder of the enzyme-bound template is widely unperturbed, relative to its fully natural counterpart.²² In contrast, at least minor distortions are apparent throughout the primer (Figure 4A).

Examination of the polymerase reveals that, relative to its open form, the presence of the unnatural base pair at the postinsertion site induces the thumb domain to rotate, with helices H, I, and K, which interact with the 3' side of the primer, moving closer to the active site, and helices H1 and H2, which interact with the 5' side of the primer, moving away (Figure 4B). The position of the fingers domain is less perturbed. The intercalated state appears to be accommodated by a network of protein residues of the fingers domain, including Asn750, Tyr671, Gln754, and Glu615, which pack on the free 3' face of the d5SICS nucleobase (Figure S3), and the primer terminus appears further stabilized by H-bonds between the 3' OH and phosphate backbone with His784 and Arg587, respectively. While the latter interaction is also observed with a natural substrate,²² the former is not (by analogy to the homologous large fragment of polymerase I from *Bacillus stearothermophilus* (Bacillus fragment, BF), the 3' OH of a fully natural substrate forms an H-bond with Asp785³⁵). Furthermore, the sulfur atom of d5SICS engages in a water mediated H-bond with Thr571. The phosphate of the template dNaM interacts with Arg746 (Figure S3), and an analogous interaction is observed with a fully natural substrate.

In the structure of $\text{KTQ}(\text{E2})_{\text{dNaM-d5SICS}}$, the unnatural base pair again forms via intercalation, but in this case, by the primer d5SICS nucleobase inserting between the template dNaM and its 3' dG ($\text{dG}_{\text{n-2}}$) (Figure 3). The extent of intercalation appears somewhat less than in $\text{KTQ}(\text{E1})_{\text{dNaM-d5SICS}}$, with a C1'–C1' distance of 9.4 Å (compared to 8.4 Å). The overall structure of the primer/template is similar to that observed with $\text{KTQ}(\text{E1})_{\text{dNaM-d5SICS}}$, but despite the decreased intercalation, it is somewhat more distorted, with the C1' of the primer unnatural nucleotide moving 5.8 Å toward the template and the C1' of the unnatural nucleotide in the template shifting 5.4 Å in the direction of translocation, relative to their positions observed with natural substrates (Figure 4A and D). In addition, dNaM shields the templating nucleobase from interacting with the primer strand, and possibly as a result, neither the downstream nucleotides nor Arg587 are well resolved in the $\text{KTQ}(\text{E2})_{\text{dNaM-d5SICS}}$ structure. While the overall structure of the polymerase is similar in $\text{KTQ}(\text{E2})_{\text{dNaM-d5SICS}}$ and $\text{KTQ}(\text{E2})_{\text{d5SICS-dNaM}}$

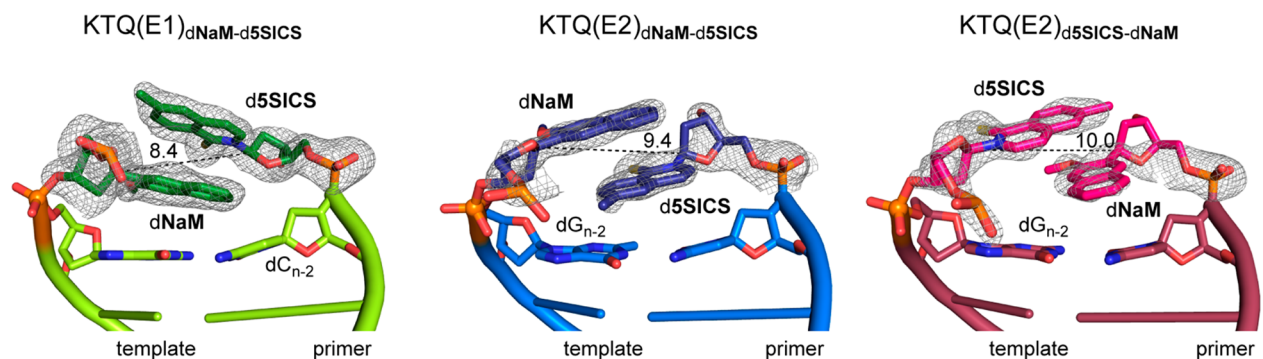


Figure 3. Primer/template arrangement of open binary complexes with dNaM-d5SICS in the postinsertion site. $\text{KTQ}(\text{E1})_{\text{dNaM-d5SICS}}$, $\text{KTQ}(\text{E2})_{\text{dNaM-d5SICS}}$, and $\text{KTQ}(\text{E2})_{\text{d5SICS-dNaM}}$ are labeled and shown in green, blue, and red, respectively. The intercalated unnatural base pair is shown in dark green, dark blue, and pink, respectively, surrounded by their simulated annealing mFo-DFc omit maps contoured at 3σ . C1'–C1' distances (Å) within each unnatural pair are shown.

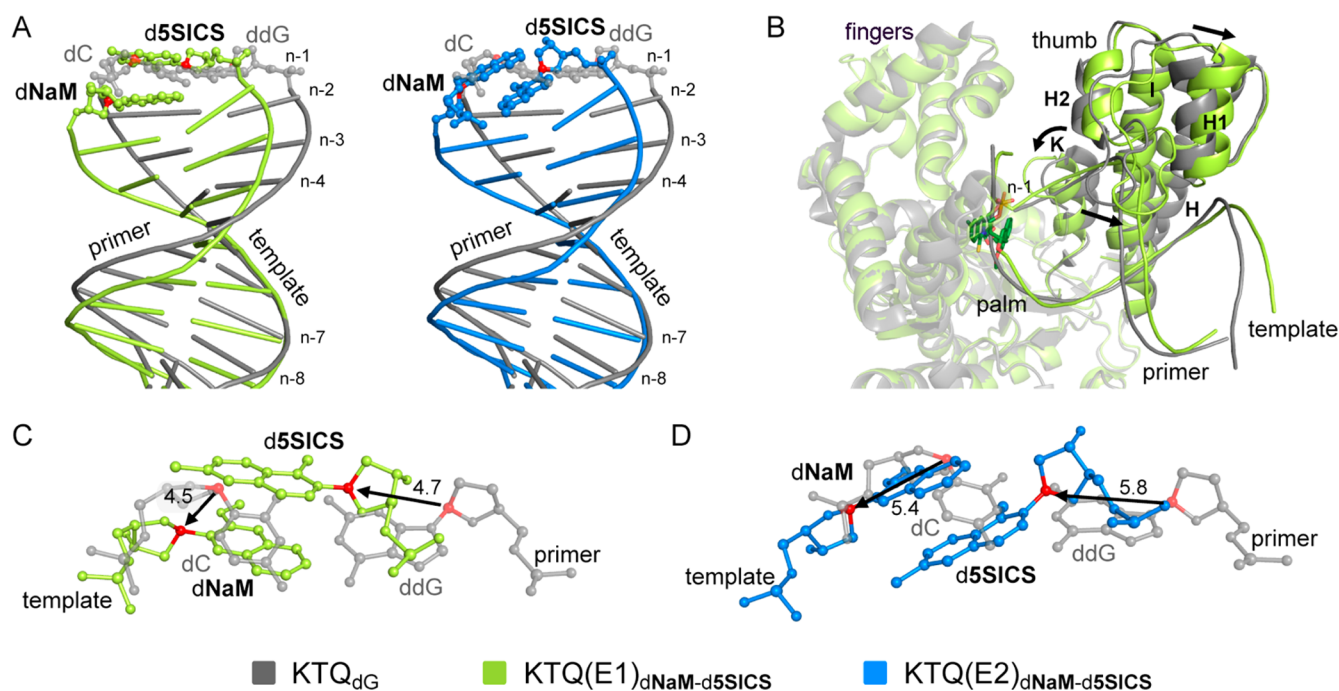


Figure 4. Comparison of KTQ(E1)_{dNaM-d5SICS} (green) and KTQ(E2)_{dNaM-d5SICS} (blue) with KTQ_{dG} with a natural dC-ddG base pair at the postinsertion site (gray). (A) Superposition of duplex portion of primer/template of KTQ(E1)_{dNaM-d5SICS} and KTQ(E2)_{dNaM-d5SICS} with KTQ_{dG}. The unnatural base pair in the postinsertion site is shown in ball and stick representation with the C1' atoms shown in red. (B) Superposition of KTQ(E1)_{dNaM-d5SICS} and KTQ_{dG}, shown as cartoon. The fingers and palm domains which are only slightly affected by the unnatural base pair are transparent. The larger movement of the thumb domain is indicated with black arrows. (C and D) Superposition of unnatural and natural base pair (from KTQ_{dG}) at the postinsertion site with distance between C1' atoms indicated in Å, (C) KTQ(E1)_{dNaM-d5SICS} and (D) KTQ(E2)_{dNaM-d5SICS}.

(E1)_{dNaM-d5SICS}, there are significant differences in the interactions with the unnatural base pair. In KTQ-(E2)_{dNaM-d5SICS}, the O-helix residue Tyr671 stacks on the template dNaM nucleobase, while Gln754 and Glu615 can form H-bonds with the sulfur of d5SICS and the primer 3'OH, respectively (Figure S3). In contrast to KTQ(E1)_{dNaM-d5SICS}, no specific interactions are observed between KlenTaq and the phosphates of the unnatural nucleotides in either the primer or template strands.

To examine the effect of strand context, we solved the structure of the KTQ(E2)_{d5SICS-dNaM} binary complex, with dNaM at the primer terminus paired opposite d5SICS in the template (Table 1). Again, the unnatural nucleobases pair in an intercalated fashion, and in this case in a manner similar to KTQ(E2)_{dNaM-d5SICS} (Figure 3), although the C1'-C1' distance of 10.0 Å is somewhat longer. The protein-DNA interactions, including those involving the unnatural base pair, are also conserved in the two structures, including the H-bond between the Gln754 H-bond donor and the H-bond acceptor of the unnatural nucleotide at the primer terminus, which with KTQ(E2)_{d5SICS-dNaM} is the methoxy group of dNaM and with KTQ(E2)_{dNaM-d5SICS} is the sulfur atom of d5SICS (Figure S3).

To demonstrate that the length of the single-stranded portion of the template does not affect the structure of the unnatural base pair at the primer terminus, we examined primer/template E3, which like E1 has the template sequence 3'-dGNaMG, but like E2 it has a 3-nt overhang (Table 1). With KTQ(E3)_{dNaM-d5SICS}, the unnatural base pair is again intercalated, and in a manner similar to that observed with KTQ(E1)_{dNaM-d5SICS} (Figure S3). Furthermore, while the last single-stranded residue of KTQ(E3)_{dNaM-d5SICS} (dT_{n+2}) is not resolved, the overall structures of KTQ(E3)_{dNaM-d5SICS} and

KTQ(E1)_{dNaM-d5SICS} are almost identical with an rmsd of 0.21 Å, and with the same residues interacting with the intercalated unnatural base pair.

Multiple attempts were made to solve the structures of the ternary complexes of either KTQ_{dNaM-d5SICS} or KTQ_{d5SICS-dNaM} and nonhydrolyzable variants of the next correct natural triphosphate, dCTP or dGTP (e.g., NHdCTP or NHdGTP). In no case were we able to detect electron density associated with the nucleoside triphosphate. Thus, we conclude that the low affinity of the natural triphosphates for the primer terminus containing an intercalated unnatural base pair precludes their crystallization in a ternary complex.

DISCUSSION

Replication of natural DNA is mediated by the H-bonding and shape complementarity of the pairing nucleobases.³⁶⁻³⁹ However, d5SICS and dNaM cannot form H-bonds and have shapes that are very different from the natural purines and pyrimidines. Nonetheless, during PCR amplification, d5SICS-dNaM is functionally equivalent to a natural base pair.¹⁹ Our earliest efforts to understand this efficient replication focused predominantly on structure-activity relationships derived from kinetics assays. Later, we focused on free duplex DNA and showed that the pair forms via cross-strand intercalation,^{22,23} raising more questions than we answered. This situation was at least partially clarified with our previous characterization of the prechemistry complexes leading to the insertion of d5SICSTP opposite dNaM. The structures of KTQ_{dNaM} and KTQ_{dNaM-d5SICSTP} elucidated a mutual induced fit mechanism, wherein pairing of d5SICS-dNaM drives the open-to-closed conformational transition of the polymerase, and the closed conformation of the polymerase induces d5SICS-dNaM to

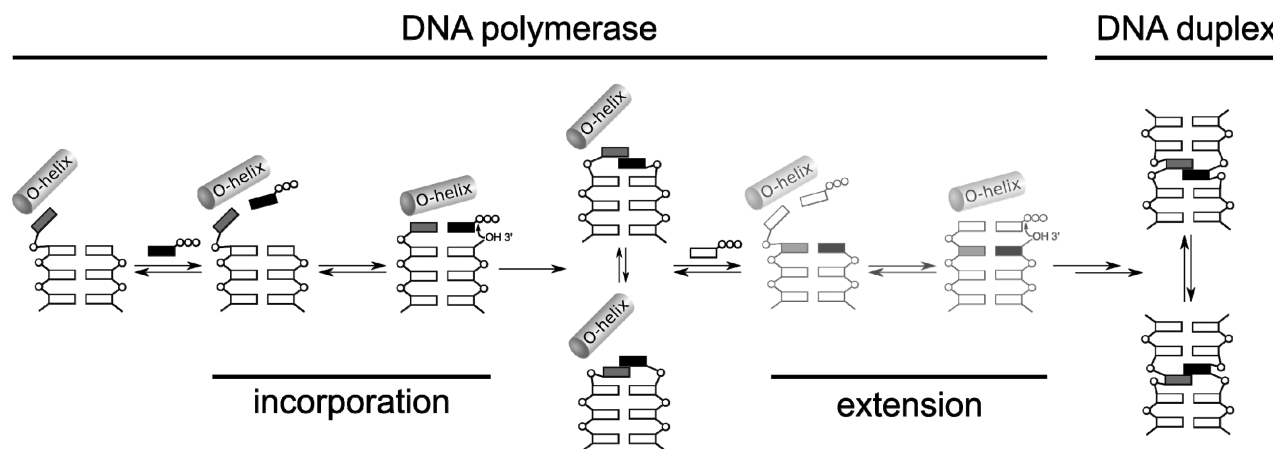


Figure 5. Proposed mechanism of replication. Intermediates not yet validated by structural studies (i.e., extension complexes) are shown in lighter color. The steps corresponding to incorporation of the unnatural triphosphate and subsequent extension of the nascent unnatural base pair are indicated. The O-helix of the protein is shown, phosphates are indicated with open circles, natural nucleotides are indicated with open rectangles, and the unnatural nucleotides are indicated with grey and black rectangles.

adopt a Watson–Crick-like structure. With these results, our attention turned to the mechanisms underlying the remaining steps of replication, including the insertion of dNaMTP opposite dSSICS, and the subsequent continued primer elongation after incorporation of either unnatural triphosphate.

Unlike with the addition of dSSICSTP to KTQ_{dNaM} ,²² the addition of dNaMTP to $\text{KTQ}_{\text{dSSICS}}$ did not induce the canonical open-to-closed conformational change observed during the synthesis of a natural base pair, but rather resulted in the formation of a structure wherein the DNA polymerase fingers domain remains in a partially open conformation and the dNaMTP is bound via its triphosphate moiety to the O-helix. A similar conformation has been described by Beese and Wu for BF polymerase with a dG-dTTP or a dG-ddTTP mismatch.⁴⁰ In this structure, the polymerase remains in a partially open conformation, referred to as “ajar,” and the templating nucleotide displaces the “gate keeping” residue Y714 (Y671 in KlenTaq) from the insertion site. It has been suggested that this ajar conformation allows the DNA polymerase to test for complementarity between the incoming and templating nucleotides before the enzyme transitions to the closed catalytically competent state. While Y761 remains in the templating position in $\text{KTQ}_{\text{dSSICS-dNaMTP}}$, both it and the nucleobase of dSSICS appear strained toward the same switch observed in the BF structure. A similar configuration has been observed with KlenTaq with an abasic site at the templating position.^{41,42} Thus, the $\text{KTQ}_{\text{dSSICS-dNaMTP}}$ complex appears trapped in an intermediate state between the open binary complex and the ajar state observed with BF, similar to a partially closed state observed with the homologous *E. coli* polymerase I via biophysical studies.^{43,44} Regardless, it is clear that incorporation of dNaMTP would require significant rearrangement of the polymerase to reach the catalytically competent closed state, while $\text{KTQ}_{\text{dNaM-dSSICSTP}}$ spontaneously forms the catalytically competent closed complex. This difference likely explains why the insertion of dNaMTP opposite dSSICS is often less efficient than the insertion of dSSICSTP opposite dNaM.¹⁶

In all four postincorporation complexes characterized, the nucleobases pair in an intercalated manner, similar to their pairing in free duplex DNA.²³ However, two modes of intercalation are observed. With primer/template complexes

E1 and E3, a common mode of intercalation is observed (Figures 3 and S3), which demonstrates that the mode of pairing is unlikely to depend on the length of the single stranded template. In this mode of intercalation, the template dNaM inserts between its pairing dSSICS and the flanking dC_{n-2} , which allows for the template dG_n to form stabilizing interactions with the primer terminus (Figure S4). In contrast, in both complexes with primer/template E2, dC_n is unable to mediate such interactions, and the intercalated structure is formed by insertion of the primer dSSICS ($\text{KTQ}(\text{E2})_{\text{dNaM-dSSICS}}$) or dNaM ($\text{KTQ}(\text{E2})_{\text{dSSICS-dNaM}}$) between its pairing unnatural nucleobase and its flanking dG_{n-2} of the template, which likely optimizes packing interactions. Surprisingly, the mode of intercalation appears to depend most on sequence-specific interactions of the flanking nucleotides, with the specific packing interactions between the intercalating nucleobases being of secondary importance.

Interestingly, the polymerase appears to be able to provide unique stabilizing interactions to the two types of intercalated structures at the primer terminus. In the $\text{KTQ}(\text{E1})_{\text{dNaM-dSSICS}}$ and $\text{KTQ}(\text{E3})_{\text{dNaM-dSSICS}}$ structures, the observed intercalated state leaves one face of the primer dSSICS unpacked by a flanking nucleobase, and its position is stabilized by packing interactions with Asn750, Tyr671, Gln754 and Glu615, an ionic interaction between its phosphate and Arg587, a water-mediated H-bond between its sulfur and Thr571, and by an H-bond between its 3'OH and His784. The position of dNaM in the template is stabilized by an ionic interaction between its phosphate and Arg746. In the $\text{KTQ}(\text{E2})_{\text{dNaM-dSSICS}}$ and $\text{KTQ}(\text{E2})_{\text{dSSICS-dNaM}}$ structures, the intercalated state adopted leaves one face of the unnatural nucleobase in the template (dNaM and dSSICS, respectively) unpacked by a flanking nucleobase, and its position is stabilized by packing interactions with O-helix residue Tyr671. In this case, the position of the primer terminus can be stabilized by an H-bond between its 3'OH and Gln615 and by an H-bond between Gln754 and the H-bond acceptor *ortho* to the glycosidic linkage (methoxy in dNaM and sulfur in dSSICS). Unlike with E1 and E3, neither the template nor primer strand in either complex with E2 is stabilized via interactions with their backbone phosphates. The rather different interactions by which the two intercalated structures are accommodated reveals that the polymerase is

surprisingly plastic. Regardless, both structures of the post-insertion complexes require deintercalation and significant remodeling of the polymerase active site for incorporation of the next dNTP, likely explaining why structures with the next correct natural triphosphate could not be obtained and also why extension of the unnatural base pair is less efficient than incorporation of the unnatural triphosphate.

Based on this and previously reported structural data,^{22,23,32,36,39,40} we propose the following mechanism of replication (Figure 5). The unnatural triphosphate initially binds to the O-helix, producing a flexible complex that samples different conformations, and when sufficiently stabilizing hydrophobic and packing interactions are made, the open-to-closed transition is induced, which induces the unnatural base pair to adopt a planar, Watson–Crick-like pairing, and incorporation of the triphosphate onto the growing primer terminus. With d5SICSTP incorporation, the intermediate states are populated only transiently, and the closed complex may only be captured by preventing incorporation with a dideoxy primer terminus. However, with dNaMTP incorporation, the series of conformational changes are halted at an ajar-like state with the unnatural triphosphate remaining bound to the O-helix, due to either the stability of this complex or the instability of the corresponding closed complex, and further progress toward the incorporation of dNaMTP requires thermal fluctuations to populate the closed state. After incorporation of either d5SICSTP or dNaMTP, the polymerase returns to the open conformation and pyrophosphate is released.⁴⁵ However, in this state, the unnatural base pair adopts a cross-strand intercalated structure, similar to the structure it adopts in free duplex DNA, and continued primer elongation requires thermal fluctuations to both deintercalate the unnatural base pair and reorganize the polymerase active site. Because extension consistently limits the replication of DNA containing the unnatural base pair, the model predicts that further optimization of d5SICS-dNaM may be possible by making changes to the nucleobase analogues that decrease the stability of the intercalated structures. Efforts to test this hypothesis are currently underway.

MATERIALS AND METHODS

Oligonucleotide Synthesis. Natural oligonucleotides were purchased from IDT (San Diego, CA). dNaM and d5SICS phosphoramidites and nucleosides were obtained from Berry & Associates Inc. (Dexter, MI), and the latter were phosphorylated using Ludwig and Eckstein conditions⁴⁶ as described.¹⁶ Oligonucleotides containing an unnatural nucleotide were prepared using standard automated DNA synthesis methodology with ultramild DNA synthesis phosphoramidites on CPG ultramild supports (1 μ mol, Glen Research; Sterling, VA) and an ABI Expedite 8905 synthesizer. After automated synthesis, the DMT-ON oligonucleotide was first purified by Glen-Pak cartridge (Glen Research) and then by 8 M urea 20% PAGE, followed by Synergi Fusion-RP HPLC (Phenomenex, Torrance, CA) to single-band purity (>98%) using a linear gradient of 100 mM triethylammonium bicarbonate buffer (pH 7.5) and acetonitrile (5–30% over 35 min). The fractions containing purified oligonucleotides were collected and dried by vacuum centrifugation, and their identity was confirmed by MALDI-ToF with THAP matrix.

Protein Production, Crystallization, and Structure Determination. KlenTaq was prepared using an *E. coli* codon-optimized gene encoding amino acids 293–832 of Taq polymerase (purchased from GeneArt, Germany) cloned into the vector pGDR11 and expressed in *E. coli* strain BL21 (DE3) in LB medium for 4 h after induction with 1 mM IPTG. The harvested cell pellet was resuspended in lysis buffer (50 mM Tris HCl pH 8.5, 10 mM MgCl₂, 16 mM (NH₄)₂SO₄ 0.1%

TritonX-100, 0.1% hydroxypolyethoxydodecane, and 1 mM PMSF) and lysed by the addition of 0.5 mg/mL lysozyme and incubation for 1 h at 37 °C. After lysis, a heat denaturation was performed (20 min, 80 °C) and the cell debris was pelleted by ultracentrifugation (1 h, 35 000g). Bacterial DNA in the supernatant was removed by PEI-precipitation and centrifugation. The resulting supernatant was purified by anion exchange chromatography (Q Sepharose) in 20 mM Tris HCl pH 8.5, 1 mM EDTA, 1 mM β -mercaptoethanol, eluting with a NaCl gradient. Fractions containing KlenTaq were pooled, concentrated, and further purified by size-exclusion chromatography (Superdex 75) in 20 mM Tris HCl pH 7.5, 1 mM EDTA, 1 mM β -mercaptoethanol, 0.15 M NaCl.

Purified KlenTaq was stored at 4 °C. Primers and templates were annealed prior to addition of protein and triphosphates. KlenTaq was mixed with triphosphate and/or primer/template and incubated for 30 min at 30 °C. The mixture was then filtered, and crystallization conditions were screened using the sitting drop vapor diffusion method at 18 °C. Hits were reproduced using either the sitting or hanging drop vapor diffusion method. Prior to measurement, crystals were flash frozen in liquid nitrogen either with or without cryo protection (see the Supporting Information).

Data was collected at the beamline PXIII (XO6DA) and PXI (XO6SA) at the Swiss Light Source of the Paul Scherrer Institute in Villigen, Switzerland. Data reduction was performed with the XDS package.⁴⁷ Statistics of data collection and refinement for all structures are given in Table S1. Data was used in refinement up to a resolution with a CC_{1/2} value⁴⁸ of around 50%. To facilitate comparison with other deposited structures, we also report resolution values at which $1/\sigma = 2$ (see Table S1). Data reduction of the KTQ_{d5SICS} and KTQ_{d5SICS-dNaMTP} data was done in space group P3₁21 (for cell dimensions, see Table S1), and the structures were solved by rigid-body refinement using a previously published KlenTaq structure (PDB: 3M8S⁴⁹) as a model. All binary elongation complexes (KTQ(E1), KTQ(E2), and KTQ(E3)) crystallized in space group C222₁ with similar cell dimensions (see Table S1). The KTQ(E1) complex was solved by molecular replacement using the binary KlenTaq structure 3SZ2²² as a search model. The KTQ(E2) and KTQ(E3) structures were solved by rigid body refinement against KTQ(E1). All structures were improved by refinement in PHENIX⁵⁰ and model building in COOT.⁵¹ During refinement, structures were evaluated using the MolProbity server.⁵² Restraint files of dNaM, d5SICS, and dNaMTP for refinement were created using the grade Web Server.⁵³ Figures were created with PyMOL.⁵⁴

ASSOCIATED CONTENT

Supporting Information

Supporting table and figures, crystallization conditions, and oligonucleotide sequences. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We thank the beamline staff of the Swiss Light Source at the Paul Scherrer Institute for their assistance during data collection. We thank Dr. Phillip Ordoukhanian for assistance with oligonucleotide synthesis. This work was supported by the National Institutes of Health (GM 060005 to F.E.R.) and by the Konstanz Research School Chemical Biology.

■ REFERENCES

- (1) Collins, M. L.; Irvine, B.; Tyner, D.; Fine, E.; Zayati, C.; Chang, C.; Horn, T.; Ahle, D.; Detmer, J.; Shen, L. P.; Kolberg, J.; Bushnell, S.; Urdea, M. S.; Ho, D. D. *Nucleic Acids Res.* **1997**, *25*, 2979–2984.
- (2) Johnson, S. C.; Marshall, D. J.; Harms, G.; Miller, C. M.; Sherrill, C. B.; Beaty, E. L.; Lederer, S. A.; Roesch, E. B.; Madsen, G.; Hoffman, G. L.; Laessig, R. H.; Kopish, G. J.; Baker, M. W.; Benner, S. A.; Farrell, P. M.; Prudent, J. R. *Clin. Chem.* **2004**, *50*, 2019–2027.
- (3) Arens, M. Q.; Buller, R. S.; Rankin, A.; Mason, S.; Whetsell, A.; Agapov, E.; Lee, W. M.; Storch, G. A. *J. Clin. Microbiol.* **2010**, *48*, 2387–2395.
- (4) Lee, W. M.; Grindle, K.; Pappas, T.; Marshall, D. J.; Moser, M. J.; Beaty, E. L.; Shult, P. A.; Prudent, J. R.; Gern, J. E. *J. Clin. Microbiol.* **2007**, *45*, 2626–2634.
- (5) Kimoto, M.; Yamashige, R.; Matsunaga, K.; Yokoyama, S.; Hirao, I. *Nat. Biotechnol.* **2013**, *31*, 453–457.
- (6) Hollenstein, M.; Hipolito, C. J.; Lam, C. H.; Perrin, D. M. *Nucleic Acids Res.* **2009**, *37*, 1638–1649.
- (7) Keefe, A. D.; Cload, S. T. *Curr. Opin. Chem. Biol.* **2008**, *12*, 448–456.
- (8) Seeman, N. C. *Annu. Rev. Biochem.* **2010**, *79*, 65–87.
- (9) Wang, H.; Yang, R.; Yang, L.; Tan, W. *ACS Nano* **2009**, *3*, 2451–2460.
- (10) Chen, T.; Shukoor, M. I.; Chen, Y.; Yuan, Q.; Zhu, Z.; Zhao, Z.; Gulbakan, B.; Tan, W. *Nanoscale* **2011**, *3*, 546–556.
- (11) Chelliserrykattil, J.; Lu, H.; Lee, A. H.; Kool, E. T. *ChemBioChem* **2008**, *9*, 2976–2980.
- (12) Brotschi, C.; Mathis, G.; Leumann, C. J. *Chem.—Eur. J.* **2005**, *11*, 1911–1923.
- (13) Kaul, C.; Muller, M.; Wagner, M.; Schneider, S.; Carell, T. *Nat. Chem.* **2011**, *3*, 794–800.
- (14) Meggers, E.; Holland, P. L.; Tolman, W. B.; Romesberg, F. E.; Schultz, P. G. *J. Am. Chem. Soc.* **2000**, *122*, 10714–10715.
- (15) Minakawa, N.; Ogata, S.; Takahashi, M.; Matsuda, A. *J. Am. Chem. Soc.* **2009**, *131*, 1644–1645.
- (16) Lavergne, T.; Malyshev, D. A.; Romesberg, F. E. *Chem.—Eur. J.* **2012**, *18*, 1231–1239.
- (17) Yamashige, R.; Kimoto, M.; Takezawa, Y.; Sato, A.; Mitsui, T.; Yokoyama, S.; Hirao, I. *Nucleic Acids Res.* **2012**, *40*, 2793–2806.
- (18) Yang, Z.; Chen, F.; Alvarado, J. B.; Benner, S. A. *J. Am. Chem. Soc.* **2011**, *133*, 15105–15112.
- (19) Malyshev, D. A.; Dhami, K.; Quach, H. T.; Lavergne, T.; Ordoukhanian, P.; Torkamani, A.; Romesberg, F. E. *Proc. Natl. Acad. Sci. U.S.A.* **2012**, *109*, 12005–12010.
- (20) Seo, Y. J.; Matsuda, S.; Romesberg, F. E. *J. Am. Chem. Soc.* **2009**, *131*, 5046–5047.
- (21) Seo, Y. J.; Malyshev, D. A.; Lavergne, T.; Ordoukhanian, P.; Romesberg, F. E. *J. Am. Chem. Soc.* **2011**, *133*, 19878–19888.
- (22) Betz, K.; Malyshev, D. A.; Lavergne, T.; Welte, W.; Diederichs, K.; Dwyer, T. J.; Ordoukhanian, P.; Romesberg, F. E.; Marx, A. *Nat. Chem. Biol.* **2012**, *8*, 612–614.
- (23) Malyshev, D. A.; Pfaff, D. A.; Ippoliti, S. I.; Hwang, G. T.; Dwyer, T. J.; Romesberg, F. E. *Chem.—Eur. J.* **2010**, *16*, 12650–12659.
- (24) Johar, Z.; Zahn, A.; Leumann, C. J.; Jaun, B. *Chem.—Eur. J.* **2008**, *14*, 1080–1086.
- (25) Matsuda, S.; Fillo, J. D.; Henry, A. A.; Rai, P.; Wilkens, S. J.; Dwyer, T. J.; Geierstanger, B. H.; Wemmer, D. E.; Schultz, P. G.; Spraggon, G.; Romesberg, F. E. *J. Am. Chem. Soc.* **2007**, *129*, 10466–10473.
- (26) Krahn, J. M.; Beard, W. A.; Wilson, S. H. *Structure* **2004**, *12*, 1823–1832.
- (27) Chou, S. H.; Chin, K. H.; Wang, A. H. *Nucleic Acids Res.* **2003**, *31*, 2461–2474.
- (28) Chou, S. H.; Zhu, L.; Reid, B. R. *J. Mol. Biol.* **1994**, *244*, 259–268.
- (29) Shepard, W.; Cruse, W. B.; Fourme, R.; de la Fortelle, E.; Prange, T. *Structure* **1998**, *6*, 849–861.
- (30) Špacková, N. A.; Berger, I.; Šponer, J. *J. Am. Chem. Soc.* **2000**, *122*, 7564–7572.
- (31) Sunami, T.; Kondo, J.; Hirao, I.; Watanabe, K.; Miura, K. I.; Takenaka, A. *Acta Crystallogr., Sect. D: Biol. Crystallogr.* **2004**, *60*, 90–96.
- (32) Rothwell, P. J.; Waksman, G. *Adv. Protein Chem.* **2005**, *71*, 401–440.
- (33) Li, Y.; Korolev, S.; Waksman, G. *EMBO J.* **1998**, *17*, 7514–7525.
- (34) Doublié, S.; Tabor, S.; Long, A. M.; Richardson, C. C.; Ellenberger, T. *Nature* **1998**, *391*, 251–258.
- (35) Kiefer, J. R.; Mao, C.; Braman, J. C.; Beese, L. S. *Nature* **1998**, *391*, 304–307.
- (36) Echols, H.; Goodman, M. F. *Annu. Rev. Biochem.* **1991**, *60*, 477–511.
- (37) Kool, E. T. *Annu. Rev. Biochem.* **2002**, *71*, 191–219.
- (38) Goodman, M. F. *Proc. Natl. Acad. Sci. U.S.A.* **1997**, *94*, 10493–10495.
- (39) Kunkel, T. A. *J. Biol. Chem.* **2004**, *279*, 16895–16898.
- (40) Wu, E. Y.; Beese, L. S. *J. Biol. Chem.* **2011**, *286*, 19758–19767.
- (41) Obeid, S.; Welte, W.; Diederichs, K.; Marx, A. *J. Biol. Chem.* **2012**, *287*, 14099–14108.
- (42) Obeid, S.; Blattner, N.; Kranaster, R.; Schnur, A.; Diederichs, K.; Welte, W.; Marx, A. *EMBO J.* **2010**, *29*, 1738–1747.
- (43) Hohlbein, J.; Aigrain, L.; Craggs, T. D.; Bermek, O.; Potapova, O.; Shoolizadeh, P.; Grindley, N. D. F.; Joyce, C. M.; Kapanidis, A. N. *Nat. Commun.* **2013**, *4*, 2131.
- (44) Berezna, S. Y.; Gill, J. P.; Lamichhane, R.; Millar, D. P. *J. Am. Chem. Soc.* **2012**, *134*, 11261–11268.
- (45) Golosov, A. A.; Warren, J. J.; Beese, L. S.; Karplus, M. *Structure* **2010**, *18*, 83–93.
- (46) Ludwig, J.; Eckstein, F. *J. Org. Chem.* **1989**, *54*, 631–635.
- (47) Kabsch, W. *Acta Crystallogr., Sect. D: Biol. Crystallogr.* **2010**, *66*, 125–132.
- (48) Karplus, P. A.; Diederichs, K. *Science* **2012**, *336*, 1030–1033.
- (49) Betz, K.; Streckenbach, F.; Schnur, A.; Exner, T.; Welte, W.; Diederichs, K.; Marx, A. *Angew. Chem., Int. Ed.* **2010**, *49*, 5181–5184.
- (50) Adams, P. D.; Afonine, P. V.; Bunkoczi, G.; Chen, V. B.; Davis, I. W.; Echols, N.; Headd, J. J.; Hung, L.-W.; Kapral, G. J.; Grosse-Kunstleve, R. W.; McCoy, A. J.; Moriarty, N. W.; Oeffner, R.; Read, R. J.; Richardson, D. C.; Richardson, J. S.; Terwilliger, T. C.; Zwart, P. H. *Acta Crystallogr., Sect. D: Biol. Crystallogr.* **2010**, *66*, 213–221.
- (51) Emsley, P.; Lohkamp, B.; Scott, W. G.; Cowtan, K. *Acta Crystallogr., Sect. D: Biol. Crystallogr.* **2010**, *66*, 486–501.
- (52) Chen, V. B.; Arendall, W. B., 3rd; Headd, J. J.; Keedy, D. A.; Immormino, R. M.; Kapral, G. J.; Murray, L. W.; Richardson, J. S.; Richardson, D. C. *Acta Crystallogr., Sect. D: Biol. Crystallogr.* **2010**, *66*, 12–21.
- (53) Smart, O. S.; Womack, T. O.; Sharff, A.; Flensburg, C.; Keller, P.; Paciorek, W.; Vonrhein, C.; Bricogne, G. *grade*, version 1.2.2; Global Phasing Limited: Cambridge, U.K., 2011; <http://www.globalphasing.com>.
- (54) DeLano, W. L. *The PyMOL Molecular Graphics System*; Schrodinger, LLC: San Carlos, CA, 2002; <http://www.pymol.org>.